## 2.996 Fundamentals of Advanced Energy Conversion Lecture Memo

Lecture number: 5

Date: February 18<sup>th</sup>, 2004

• Chemical thermodynamics: Adiabatic flame temperature

**Enthalpy of reaction** 

Low and high heating values

- Availability of a chemical reaction: Gibbs free energy, efficiency
- Direct conversion: Fuel cell

Efficiency change with respect to operating temperature

**Reactions in fuel cells** 

**Electrolyzer** 

• Indirect conversion: Carnot cycle efficiency

**Internal combustion engine (SI)** 

Otto cycle efficiencies

- 1) Air standard cycle
- 2) Fuel/air cycle
- 3) Complete/incomplete combustions