#### Lecture 15: Hierarchically Ordered BCP-Nanoparticle Composites

#### **Polymer Based Nanocomposites**

Matrix: Polymers and Block Copolymers Filler: Nanoparticles

> 0, 1, 2D Fillers Ligands for Dispersion

#### **Co-assembly of BCP + Ex situ synthesized NP**

**BCP** Template for Control of Location and Orientation of NP

**Morphological Interplay and Emergent Properties** 

## Nanocomposites

- **Nanocomposites:** Heterogeneous materials with at least one characteristic length scale in the nm range
- Polymer Nanocomposites are comprised of a polymeric material and a nanoscale material.
  - -Typically made at 1-5 vol% of nanoparticles
  - New properties arising from: particle size and shape, particle locations and (possible) connectivity/proximity of particles

#### • Many factors affect polymer nanocomposite structure:

- Synthesis method (melt compounding, solvent blending, in-situ polymerization, emulsion polymerization etc).
- -Type of nanoparticles and their surface treatments (ligand shells)
- -Polymer matrix (Crystallinity, Molecular Weight, Polymer Chemistry, Blocks...)
- -Nanocomposite morphology: <u>Control of location and orientation of NP</u>

## • Understanding and optimizing composite properties is very challenging and important.

## **NanoComposite Opportunities**

Properties become **size and shape dependent** below some critical length scale.

**Dynamically tunable** materials and properties.

Hybrid material combinations *unattainable* in nature.

Sophisticated tailoring of Composite Properties

New materials, new properties, new phenomena Hierarchical structures; gradients, proximity effects...

Spatial and Orientational Ordered NP --relatively unexplored regime - lots of potential!

## **Crystalline** NanoParticles

Optical properties determined by quantum confinement effects, (and scattering, absorption, dielectric constant)

#### Metal: (plasmonics)

 $\rightarrow$  surface scattering affects electronic properties for particle size < mean free path of an electron (plasmon: coherent electron oscillation); energylevel discretization Variable size CdSe dots for size < 1.0 nm (metals become insulators!)

#### Semiconductor:

 $\rightarrow$  band gap widens for size < exciton radius (plus high photoluminescence efficiency)

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### **NanoWorld Surprises:** Mechanical Example: Polymer Nanocomposites



## **Polymer Nanocomposites**

#### **Outstanding Property Enhancements**



Figure by MIT OCW.

## Characteristics of NP Composites

- Low vol % particle-particle correlation threshold
  –Ultra low percolation threshold (~ 0.1 vol%)
- Particle number density up to ~  $10^{20}$  / cm<sup>3</sup>
- S/V per particle of ~  $10^7 \text{ cm}^2/\text{cm}^3$
- Particle size, interparticle spacing and  $R_g$  of the polymer host are all comparable

#### Conventional Composites vs. NanoComposites

Image removed due to copyright restrictions.

Please see Fig. 1 in Bockstaller, Michael M., et al. "Block Copolymer Nanocomposites: Perspectives for Tailored Materials." *Advanced Materials* 17 (2005): 1331-1349.

## **Economics of Additives**

- MMT nanoclays
- Carbon fibers
- POSS®
- MWNT (multiwalled)
- SWNT (single walled)
- n-Silica
- n-Aluminum Oxide
- n-Titanium Dioxide

- \$3.5/lb
- \$95/lb
- \$1,000/lb-R&D
- \$3,178/lb (\$7/g)-R&D
- \$227,000/lb (\$500/g)
- \$8.5/lb
- \$11.8/lb
- \$11.8/lb

## The Future - CNT the ultimate polymer?

E = 1 TPa(exp and calc)  $\sigma_{\rm f} \sim 50 + {\rm GPa}$ (calc) Values for individual tubes

#### **CNT Satellite** Tether

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Please see the cover of Scientist Today, July/August 1997.

## Single wall CNT fibres





**TEM images -** note catalyst particles in left image.

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## 2D Sheet-like NPs: Sodium Montmorillonite

- Clay
- 2D Sheet-like structure
- (2:1) phyllosilicate
- 300:1 Aspect ratio
- Isomorphoric substituted cations
- Counter ions in galleries hold individual sheets together to form 10-20 sheet stacks called "tactoids"



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Lateral dimensions > 300 nm, Individual layer thickness ~ 1nm

## **NP-Block Polymer Materials Platform**

#### **BCP Host Properties:**

relatively

#### Accessing New Properties via NP Additives

- 0, 1 and 2D particles (e.g. Q Dots, SWCNTs, Clays)
- spatial and orientational ordering of particles
- emergent properties from proximity effects

Nanoparticle Composites: MULTIFUNCTIONAL MATERIALS

## Key Attributes of NP-BCP Composites

- Size, shape, symmetry of both the NP and the BCP host
- Thermodynamic interactions (ligands and polymer)
- At least one NP dimension < one BCP domain length
- Processing conditions applied fields
- Emergent properties

## 2 Approaches to NP-BCP Composites

## (1) In situ synthesis of NP within BCP matrix

Diffusion/reaction/nucleation but restricted NP synthesis pathways since chemistry must occur inside a polymeric matrix

## (2) *Ex situ synthesis* of NP followed by blending into BCP matrix:

Synthesis is done under preferrred conditions (e.g. in solution w/o oxygen, well stirred, homogeneous reactions...)

Followed by Co-assembly of NP + BCP ~ (equilibrium thermodynamics)

#### Nanoparticle Block Copolymer Composites

Novel Microstructured Materials With Tunable Properties

#### Nanoparticle

Strong, Ex situ Synthesis Stiff, & & High dielectric constant, Co-assembly Luminescent, Magnetic, Impermeable Conductive (thermal/electrical) -basically NPs provide property enhancements not available with polymers.

#### **Block Polymer**

Self assembly in 1D, 2D, 3D Processing MultifunctionalPropertie Glassy, rubbery

#### Unique properties of the Nanocomposite arise from :

Small distances between components Confinement & Compartmentalization Ultra-large interfacial area per volume Stabilization of non-equilibrium phases Multifunctionality-tailoring of properties Size-dependent physics & chemistry

#### O-D Nanoparticles TEM GOLD SiO<sub>2</sub>

Image removed due to copyright restrictions.

Please see Supporting Material for Bockstaller, Michael R., et al. "Size-Selective Organization of Enthalpic Compatibilized Nanocrystals in Ternary Block Copolymer/Particle Mixtures." *Journal of the ACS* 125 (2003): 5276-5277.

 $\langle d \rangle = 3-5 \text{ nm}$ 

 $\langle d \rangle = 45 \text{ nm}$ 

#### Ex situ Synthesis of NP Ligands



#### Self Organization via Co-assembly of *XD-YD* NP-BCP Materials

#### **Parameters**

- Nanoparticle: size and shape: a, b, c and *IPDS* X = 0, 1, 2D dimensional
- BCP domain size and shape: A, B, C and *IMDS* Y = 1, 2, 3D periodic
- Corona: ligand chemistry, grafting density, MW of ligand
- BCP: composition, architecture, MW
- Interaction parameters:  $\chi_{ij}$

Example: 1D NP in 2D BCP

Hex packed cylinders

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Please see Fig. 2 in Bockstaller, Michael M., et al. "Block Copolymer Nanocomposites: Perspectives for Tailored Materials." *Advanced Materials* 17 (2005): 1331-1349.

nanorod

## Morphological Interplay: Co-assembly of NP-BCP Materials

- Order-Order Phase Transitions:
   Volume fraction driven (NPs increase the effective vol fraction of the microdomains that they reside in)
  - Shape accomodation driven

-e.g. NP Sheet + BCP cylinder domain -> BCP Lamellae

- -e.g. Curved NP rods + BCP cylinders -> BCP Lamellae
- NP Templating of BCP:
  - Heterogeneous nucleation of BCP domains on NP
  - Kinetics of transformation is enhanced
  - Orientational ordering of BCP by flow orienting NP
- Field Assisted Assembly: Top down <-> Bottom Up
  - Topographic confinement: commensuration to template
  - Flow, magnetic and electric fields

## NP-BCP Symmetry Compatibility Map

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Please see Fig. 4 in Bockstaller, Michael M., et al. "Block Copolymer Nanocomposites: Perspectives for Tailored Materials." *Advanced Materials* 17 (2005): 1331-1349.

## **Compatible NP-BCP Nanocomposites**

#### 0D - 3D

#### 1D - 2D

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Please see Fig. 6 in Bockstaller, Michael M., et al. "Block Copolymer Nanocomposites: Perspectives for Tailored Materials." *Advanced Materials* 17 (2005): 1331-1349.

#### ~2D - 3D

#### 2D - 1D

## Chain Topology Issues

#### **Block Polymers**

Diblock

Limited Interdigitation

#### Triblock

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Please see Fig. 5 in Bockstaller, Michael M., et al. "Block Copolymer Nanocomposites: Perspectives for Tailored Materials." *Advanced Materials* 17 (2005): 1331-1349.



#### **NP/Block Polymer Composites**

Next to NP: Loops only

## Particle-Matrix Energetics

Incorporation of NPs into the microdomains, locally deforms the chains.

#### **Chain deformation and IMDS area increase**

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Please see Fig. 3 in Bockstaller, Michael M., et al. "Block Copolymer Nanocomposites: Perspectives for Tailored Materials." *Advanced Materials* 17 (2005): 1331-1349.

## Location, Location, Location

□ Target a specific domain

Locate within the domain

- Homogeneous
- Interfacial
- Central





## **0D-1D NP-BCP** Nanocomposites

#### Location:Interfacial

#### **Small Au particles**

Gold NP

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Please see Fig. 9 in Bockstaller, Michael M., et al. "Block Copolymer Nanocomposites: Perspectives for Tailored Materials." *Advanced Materials* 17 (2005): 1331-1349.

Silica NP

#### Location:Center

Medium size SiO<sub>2</sub> particles

Ternary Nanocomposite Locations:Interfacial & Center

(see next slide for details)

# Ternary NP/BCP Nanocomposite<br/>(2 types of particles)<br/>Control of Particle LocationPS-PEP+SiO2-R2( $\phi \sim 0.04$ )+Au-S-C18H37 ( $\phi \sim 0.04$ )<br/>Cross sectional TEM

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Please see Fig. 2 in Bockstaller, Michael R., et al. "Size-Selective Organization of Enthalpic Compatibilized Nanocrystals in Ternary Block Copolymer/Particle Mixtures." *Journal of the ACS* 125 (2003): 5276-5277.

Au $\langle d \rangle = 3 \text{ nm}$ Located near the IMDSSiO2 $\langle d \rangle = 22 \text{ nm}$ Located near the domain center

## Electrical Anisotropy in Nanocomposites

Ion transport is 100 times greater parallel to clay layers than when ions have to navigate around the high aspect ratio platelets

Images removed due to copyright restrictions.

GISAXS showing anisotropic orientation of clay platelets

## Hybrid organic/inorganic systems

Please see Scheme 7 in Simon, Peter F. W., et al. "Block Copolymer-Ceramic Hybrid Materials from Organically Modified Ceramic Precursors." *Chemical Materials* 13 (2001): 3464-3468.

Mechanical properties can be tuned over several orders of magnitude !!!

The BCP is used as a "structure directing agent" for the inorganic precursor materials. A condensation reaction takes place leading to the formation of the inorganic material



Figure by MIT OCW.