1.89, Environmental Microbiology Prof. Martin Polz Lecture 20

Organic Pollutants

- Biodegradation
 - \rightarrow Mineralization: organic substrates oxidized to CO_2
 - → Partial degradation:

Co-metabolism → substrate partially degraded without energy gain Example: TCE degradation by methanotrophs → non-specific action of certain enzymes (example: oxygenases)

 \rightarrow <u>Polymerization</u>: substrates will polymerize into more complex structures

Factors Affecting Biodegradation

1) Genetic Potential

acclimation: period before onset of biodegradation

- -> organisms need to reach critical biomass
- \rightarrow favorability of biodegradation
- \rightarrow limiting nutrients
- pathways "assembled" example (selection pressure affects spread of TNTdegraders)
- 2) Bioavailability
 - \rightarrow aging
 - \rightarrow sequestration
 - \rightarrow complexation
 - Substrates are hydrophobic

 \rightarrow partition into the organics (humics) of soil

- can also become trapped in micropores of soil
- polymerization

Consequences

- o less bio available for plants and animals
- o uncertain fate and long persistence

- 3) Contaminant structure
 - a) Steric effects

branching and functional groups can hinder interactions with active sites in enzymes. Example: <u>tertiary C</u>:

Example: combinations of <u>aromatic rings with alkyl moieties</u> ring numbers in polyaromatic hydrocarbons

b) Electronic effects

<u>Functional groups</u> or residues on aromatic rings <u>can affect chemical</u> <u>properties</u> of rings because they can be:

electron donating → example: -CH₃
electron withdrawing → example: -CI
→ substituent types: Xenophores
-CI, -NO₂, -SO₃H, Br, -CN, -CF₃
→ number of substituents
→ position of substituents

4) Toxicity: must have extremely high concentrations of toxins, which can intercalate into the cell membrane and disrupt function

not as generally true as stated \rightarrow but for hydrocarbons the case

> Halogenated Hydrocarbon Degradation 3 basic mechanisms

> > \rightarrow highly dependent on number of substituents

- <u>Aerobic degradation</u> by oxygenases: works best with substrates carrying <u>few halogens</u>.
- <u>Anaerobic degradation</u> by enzymes carrying reduced metals: works best with substrates carrying <u>many halogens</u>.