## Problem Set \#4

Due: March 10, 4:00 pm

1. Provide the product of each reaction shown below. Include stereochemistry.

2. Provide the major product of each reaction.

3. Provide the best alkene starting material and reagents required to form each product.
a)

b)



d)

e)

f)

g)

4. Provide all the possible products for the following reaction. Which product is more stable and why?


HBr
5. Provide all stereoisomers that result from the following reactions. For b-f, indicate if the products are optically active (C), optically inactive/racemic (R), or optically inactive/meso (M).
a)
 $\xrightarrow[\text { 2. } \mathrm{H}_{2} \mathrm{O}_{2}, \mathrm{H}_{2} \mathrm{O}]{\text { 1. } \mathrm{O}_{3}}$
b)


Hint: build a mode!!
c)

d)

e)


f)


6. a) Draw the intermediate that results from the reaction of $\mathrm{Cl}_{2}$ with $(E)$-3-methyl-3hexene. b) Show the initial orbital overlap for the second step of this reaction. c) Explain why the nucleophile attacks the more substituted atom in the second step by drawing transition states.
7. Using what you know about the first step of ozonolysis, show the mechanism and product for the following reaction.

8. Lactones, or cyclic esters such as the starting material shown below, are prepared by halolactonization, an addition reaction to an alkene. The following lactonization is a key intermediate in the synthesis of prostaglandin PGF $_{2 \mathrm{~L}}$. Draw a stepwise mechanism for this addition reaction.

9. Show the mechanism for each reaction. Ignore stereochemistry.
a)

b)

c)

10. Show the mechanism and major product for each reaction.
a)
 HBr

$$
\longrightarrow
$$

b)

HBr

