

10.542 – Biochemical Engineering
Spring 2005

Practice Problems for Quiz #2 – Structured Models, Continuous Culture

These problems do not need to be turned in.

1) Consider a structured model that assumes the cell is comprised of four components:

- P, the concentration of intracellular precursors
- M_1 , the concentration of enzymes and RNA involved in cell synthesis
- M_2 , the concentration of structural macromolecules
- G, storage polymers

The total biomass is thus equal to the sum of the four components: $X_T = P + M_1 + M_2 + G$, where all concentrations are in g/L of reactor volume.

Develop expressions for dG/dt , dM_1/dt , dM_2/dt , and dP/dt by assuming the following:

- (i) Growth of the cell is limited by the concentration of glucose, S.
- (ii) dG/dt is equal to the rate of G formation minus the rate of G degradation. The rate of G formation per unit reactor volume obeys Michaelis-Menten kinetics in the precursor concentration per cell mass and the concentration of M_1 per unit reactor volume. The rate of G degradation is first-order in G.
- (iii) dM_2/dt obeys Michaelis-Menten kinetics in the precursor concentration per cell mass and the concentration of M_1 per unit reactor volume.
- (iv) dM_1/dt is first order in both M_1 and M_2 ; the second-order rate constant exhibits Michaelis-Menten dependence on the precursor concentration per cell mass.
- (v) dP/dt is proportional to the rate of S uptake per cell mass minus the time derivatives of M_1 , M_2 , and G.

Use the following notation:

$$k_1 = \text{g G/g } M_1\text{-hr}$$

$$k_2 = \text{g } M_2\text{/g } M_1\text{-hr}$$

$$v_s = \text{g S transported/g XT-hr}$$

$$\gamma_i = \text{stoichiometric coefficients}$$

$$k_{1D} \text{ (degradation)} = 1/\text{hr}$$

$$k_3 = \text{g } M_1\text{/g } M_2\text{-g } M_1\text{-hr}$$

$$K_i = \text{saturation parameters (Michaelis constants)}$$

2) Shuler & Kargi, Problem 6.13

3) Shuler & Kargi, Problem 6.15

4) Shuler & Kargi, Problem 6.18

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